

Tetrahedron Letters 42 (2001) 8709-8711

TETRAHEDRON LETTERS

Towards colombiasin A

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Abstract—A synthetic route to an unnatural diastereoisomer of colombiasin A is described. Key features are an arene alkylation with a γ -methylene- γ -butyrolactone and an intramolecular Diels–Alder cycloaddition. © 2001 Elsevier Science Ltd. All rights reserved.

Colombiasin A 1 is a structurally complex diterpene, recently isolated from the Caribbean sea whip *Pseudopterogorgia elisabethae.*¹ Though modest in size, its carbon frame contains three six-membered rings that share a common carbon to carbon bond. To this 'propellane' is fused a further cyclopentane ring. Six contiguous stereogenic centres are present within its unique tetracyclic skeleton, including two adjacent quaternary centres at its core (Fig. 1). Earlier this year, Nicolaou et al. reported the first total synthesis of colombiasin A.² Herein we report an alternative approach leading to an unnatural diastereoisomer of this challenging target.



Colombiasin A 1



Figure 1. Colombiane numbering.

Our synthesis began with 2,6-dimethoxytoluene 2 which was converted into phenol 4 using a known sequence.³ A Pechmann reaction with ethyl acetoacetate and mediated by triflic acid then gave coumarin 5.⁴ Reduction of the alkene and saponification of the lactone in the presence of dimethyl sulfate next furnished ester 7, a precursor to aldehyde 9. Union of aldehyde 9 and phosphonate 10 led to a separable 3:2 mixture of (*Z*)-and (*E*)-alkenes 11.⁵ Each of these materials underwent a diastereoselective aromatic alkylation on exposure to triflic acid giving spirolactone 12 as the major product. This was accompanied by traces of its diastereoisomeric partner and the cyclopentadienone 13 (Scheme 1).

At this juncture we attempted to effect a simultaneous alkene hydrogenation and ester hydrogenolysis under neutral conditions in order to establish the C-6 and C-7 stereogenic centres of colombiasin A. Unfortunately, even under forcing conditions we were unable to accomplish the second of these tasks, the product given being a single diastereoisomer of spirolactone 14 with the C-7 stereocentre installed with the correct relative stereochemistry. When the reaction was conducted in acidic media, reduction of the lactone was accomplished at room temperature and pressure, but proceeded with complete retention of configuration to give 16! This was shown to be due to the formation of γ,δ -unsaturated ester 17 as an intermediate, with the pseudo-axial C-18 methyl group dictating the stereochemical course of the reduction (Scheme 2). Reduction of spirolactone 12 to 16 could also be effected in good vield using these conditions.

Though disappointed by our failure to establish the C-6 stereogenic centre correctly, we decided to progress ester 16 towards an unnatural colombiasin diastereoisomer. Thus, it was smoothly transformed into aldehyde

0040-4039/01/\$ - see front matter @ 2001 Elsevier Science Ltd. All rights reserved. PII: S0040-4039(01)01885-8

Keywords: cyclisation; lactones; natural products; terpenes and terpenoids.

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Scheme 1.

19 allowing a diene unit to be introduced by standard Wittig chemistry. Attempts to effect aromatic oxidation of **20** to a quinone led to a complex product mixture, many components seemingly derived from oxidation of the diene. Such a problem had been overcome by Nicolaou et al.,² through protection of the diene moiety as its sulfur dioxide adduct. Their tactic proved similarly effective in our hands, though oxidation of **21** to quinone **22** with silver oxide and nitric acid was still low yielding. Finally, heating a toluene solution of **22** in the dark at 180°C in a sealed tube induced the

intramolecular Diels-Alder cycloaddition leading to a single diastereoisomer of dione 24 (Scheme 3).

At this time we must add a cautionary note on the stereochemistry of 24 as it has yet to be established with rigour. Our assignment is based on a knowledge of the stereochemistry of the precursor 22, an assumption that no epimerisation of the transient diene 23 occurs and that the Diels-Alder cycloaddition to 24 proceeds though an *endo*-cyclic transition state. All concur with the findings of Nicolaou et al. Moreover, epimerisation



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at the C-6 stereocentre in 23 would have provided us with the methyl ether of colombiasin A. Data reported for that material differs from that attained for 24, again suggesting the outcome depicted.⁶

In conclusion, we have developed a route to an unnatural diastereoisomer of colombiasin A. Key features are the arene alkylation with a γ -methylene- γ -butyrolactone, viz. 11 \rightarrow 12, and the intramolecular Diels-Alder cycloaddition 23 \rightarrow 24. We are presently seeking a more efficient protocol for effecting the aromatic oxidation 21 \rightarrow 22 and exploring other routes to 15, from which the natural colombiasin A diastereoisomer should be accessible.

Acknowledgements

The authors thank EPSRC for a Quota Studentship to M.J.T.

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- 6. Data for 24: IR (film) v_{max} 2931 m, 1673 s, 1626 w, 1457 m, 1230 w, 1127 s cm⁻¹; ¹H NMR (400 MHz, CDCl₃) $\delta_{\rm H}$ 5.67 (1H, br. s, =CH), 3.89 (3H, s, OCH₃), 3.09 (1H, m, =CHCH), 2.99 (1H, m, CCHCH₃), 2.33-1.97 (7H, m), 1.97 (3H, s, CH₃C=C-C=O), 1.44 (3H, s, CH₃C=CH), 1.44–1.10 (3H, m), 1.10 (3H, d, J 7.7, CH₃CHC), 0.94 (3H, d, J 6.9, CH₃CHCH); ¹³C NMR (100 MHz, CDCl₃) $\delta_{\rm C}$ 203.1 (s, C-14), 198.6 (s, C-17), 155.2 (s, C-16), 131.7 (s, C-11), 129.9 (s, C-15), 125.7 (d, C-10), 64.8 (s, C-1), 60.2 (q, OCH₃), 52.6 (s, C-2), 48.1 (d, C-7), 44.6 (d, C-6), 39.4 (t, C-12), 38.1 (d, C-9), 34.0 (d, C-3), 33.3 (t, C-8), 29.8 (t, C-5), 28.9 (t, C-4), 22.7 (q, C-13), 16.1 (q, C-19), 15.6 (q, C-18), 10.4 (q, C-20) assigned with reference to those data reported for 1 and colombiasin A methyl ether;^{1,2} LRMS (CI) *m*/*z* 329 (100%, MH⁺), 301 (18%, $[M-CO]^+$, 109 (42%); HRMS (EI) m/z Found M⁺: 328.2038, C₂₁H₂₈O₃ requires 328.2038.